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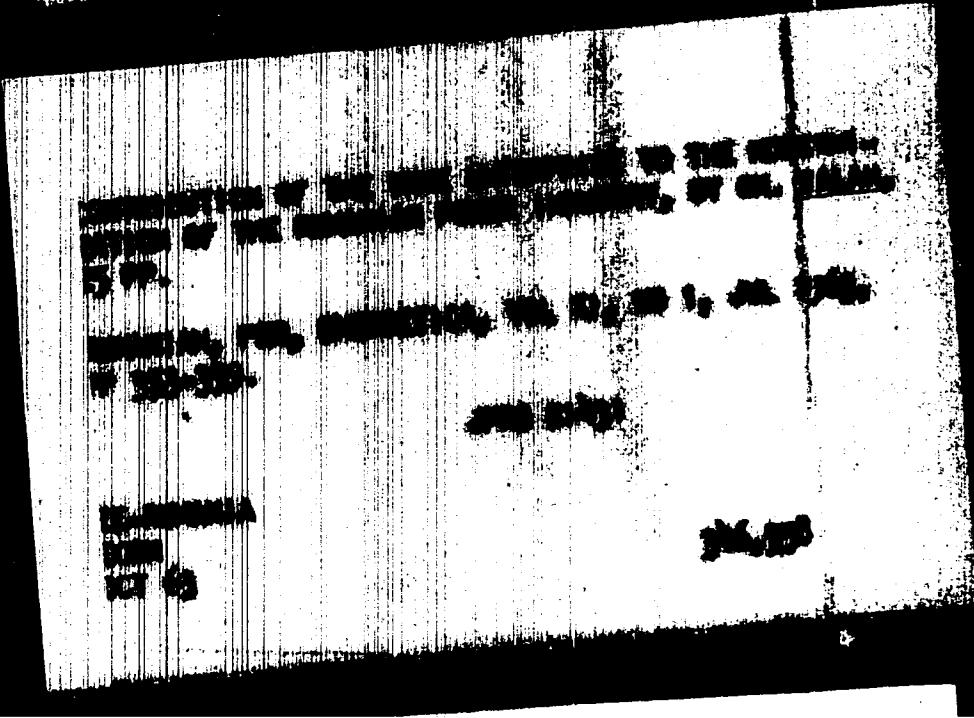
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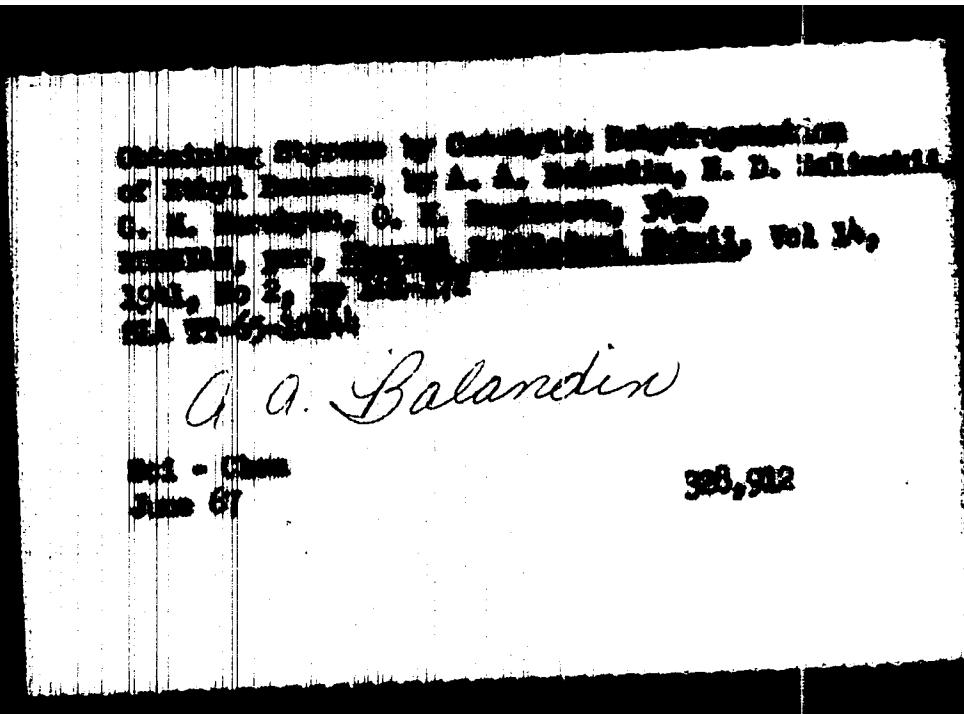
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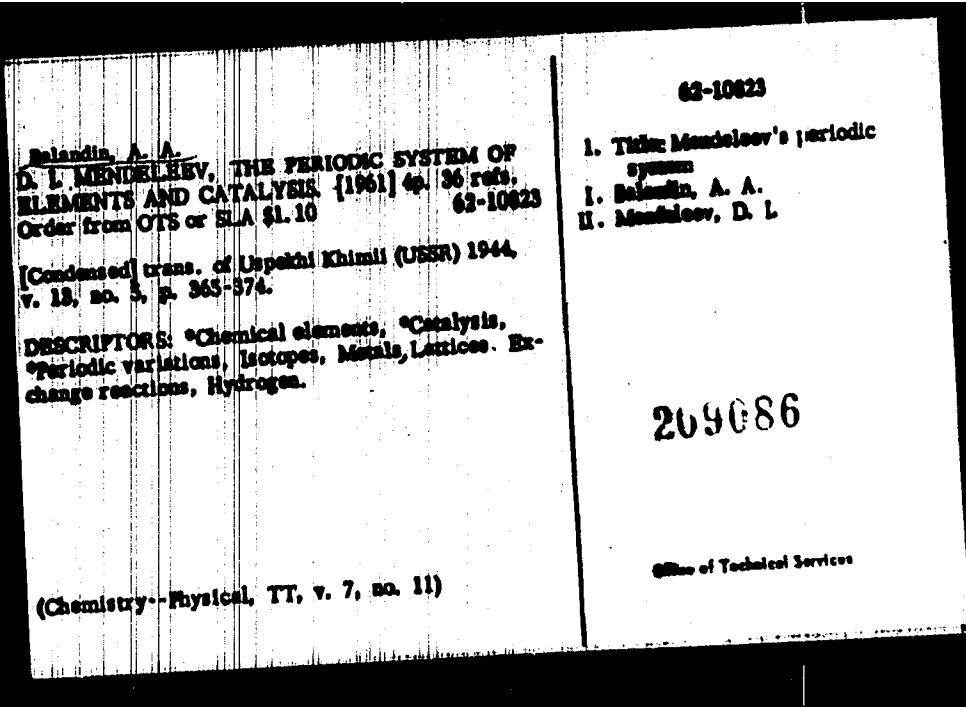
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DESCRIPTORS: *Butadienes, Synthesis (Chemistry), *Butanes, *Butenes, *Dehydrogenation, Hydrogenation, Reaction kinetics, *Labeled substances,	Under conditions of B.B. mixtures dehydrogenation the formation of butadiene occurs on the account of dehydro- genation of butylenes; butane is practically not directly converted to butadiene. In other words, butylene is the precursor of butadiene, whereby its desorption rate (Chemistry--Physical, TT, v. 11, no. 9) (over)
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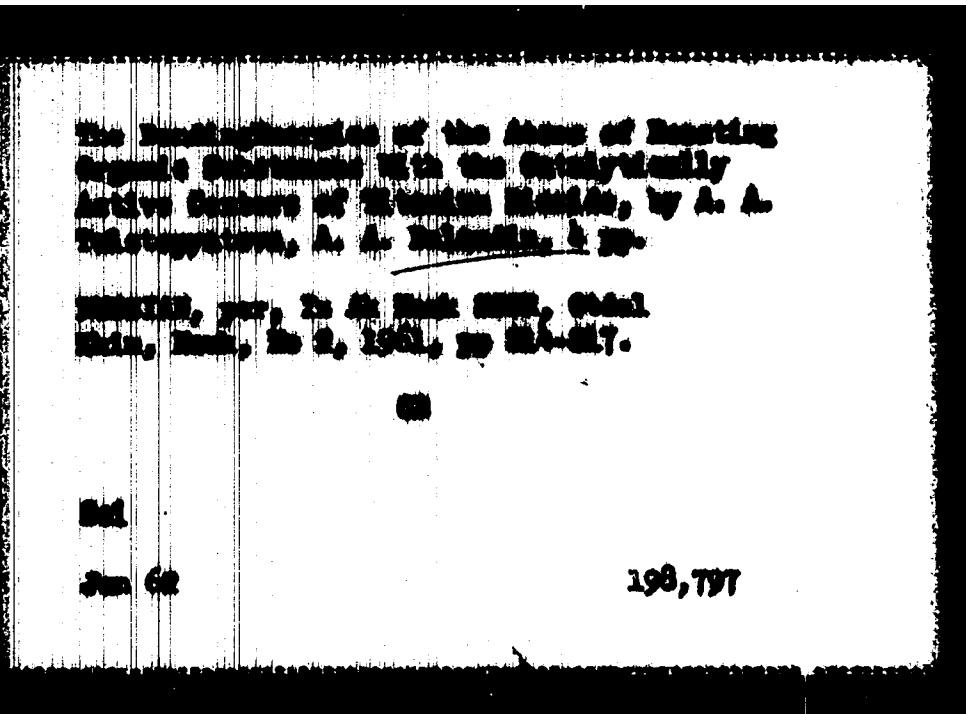
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DESCRIPTORS: *Valence, Theory, *Catalysis,
*Adsorption, Molecular association, Metals, Lattices,
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Order from ATS \$15.00 ATS-21M45R

Trans. of *Akad[emy] Nauk SSSR. Otdel[enije]
Khimicheskikh Nauk Izvestiya 1960, no. 4,
p. 614-623

- 61-12326
1. Cyclohexanols--Chemical reactions
 2. Charcoal--Chemical effects
 - I . Balandin, A. A.
 - II . Levi, G. I.
 - III . Broude, Ye. L.
 - IV . ATS-21M45R
 - V . Associated Technical Services, Inc., East Orange, N. J.

Office of Technical Services

(Chemistry--Organic, IT, v. 5, no. 5)

Balandin, A. A. and Kostin, F. K.
KINETIC STUDY OF DEHYDROGENATION OF
CYCLOHEXANE. [1941] 2p. 14 refs.
Order from OTS or SLA \$1.10

Trans. of Acta Physicochimica U R.S.S., 1942, v. 17,
p. 212-217.

61-16644
I. Balandin, A. A.
II. Kostin, F. K.

DESCRIPTORS: *Dehydrogenation, *Cyclohexanes,
Catalysts, Hydrocarbons, Benzenes, Reaction kinetics.

The catalytic dehydrogenation of cyclohexane was investigated over copper on chromium oxide, copper on chromium oxide with admixture of barium oxide and chromium oxide ex chromium chromate. The influence of temperature upon the reaction rate was studied. Benzene is the predominating resulting product. Its yield at 530° is about 100%, 13% and 51% with the first, second and third catalysts, respectively. The conclusion is reached that on the active border lines of the (Chemistry-Physical, IT, v. 6, no. 8) (over)

Office of Technical Services

Balandin, A. A.
CATALYSIS AND THE CHANGE OF MOLECULAR
FORM. [1961] 34p. 132 refs.
Order from OTS or SLA \$3.60

Trans. of Uspekhi Khimii (USSR) 1941, v. 10,
p. 262-293.

DESCRIPTORS: *Catalysis, Molecular Structure,
Hydrocarbons, Dehydrogenation, Hydrogenation,
*Molecular Isomerism, *Benzene.

An account of the researches in the application of catalysis to organic chemistry pursued by N. D. Zelinskii during 57 years up to 1941 is given, including investigations of some of his students. The investigations of his school are reviewed pertaining to the phenomenon of selective dehydrogenation leading to formation of aromatics from naphthalenes at 300°C, but leaving unaffected those naphthalenes which cannot be converted into (Chemistry--Organic, TT, v. 6, no. 8) (over)

61-16915

1. Title: Naphthalene
1. Balandin, A. A.

Office of Technical Services

Slowing Down of Surface Reactions on Catalysts and
the Mobility of Adsorbed Molecules, by V. Ye.
Vasserberg, A. A. Balandin, 4 pp.

RUSSIAN, per, Dok Ak Akad SSSR, Vol CXL, No 5, 1961,
pp 1110-1113.

CB

Sci

196-522

May 62

Reproduction of documents and photographs on file,
by Mr. A. M. Tolokonnikov, Mr. A. Tolokonnikov, T-30.

Original, year, in Mr. Tolokonnikov, Ural Rail
Bank, No. 5, 1960, no. TUT-T93.

194,610

Investigation of Catalytic Conversions of
Isopropanol and Cyclic Hydrocarbons of Titanium
Dioxide (Anatase) by Means of a Differential
Thermocouple, by A. A. Balandin, A. A.
Tolstopyatova, 6 pp.

RUSSIAN, per, Iz Ak Nauk SSSR, Otdel Khim Nauk,
No 12, 1960, pp. ~~xx~~ 2096-2102.

CB

Sci
May 62

194,490

The Formation of Carbon Dendrites in the Decomposition
of Alcohols on Nickel, by A. A. Balandin, A. P.
Rudenko, G. Stegner, Opp.

RUSSIAN, per, Iz Ak Nauk SSSR, Otdel Khim Nauk,
No 5, 1961, pp 662-770

CB

Sci
Mar 62

189,009

Determination of Relative Adsorption Coefficients by
the Isotope Dilution Method, by G. V. Isagulyants,
A. A. Bulandin, E. I. Popov, 3 pp.

RUSSIAN, per, Dok Ak Nauk SSSR, Vol CXXXIX, No 1,
1961, pp 139-141.

CB

190,303

Sci

Apr 62

Reactivity of Halogenobenzenes in Catalytic
Hydrolysis in the Vapor Phase, by L. Kh. Freidlin,
A. A. Balandin, G. A. Fridman.

RUSSIAN, per, Iz Ak Nauk SSSR, Otdel Khim Nauk,
1945, pp 655-663.

CNS 61-18988

Sci

189,552

Mar 62

Sciez Vol VIII, No 2

Catalytic Conversions of Some Alkyl Halides, by
A. A. Balandin, A. I. Kukina, I. P. Baryshnikova,
5 pp.

RUSSIAN, per, Iz Ak Nauk SSSR, Otdel Khim Nauk,
No 7, 1960, pp 1162-1169.

CB

165,862

Sci
Sep 61

Centenary of the Birth of Academician, by N. D.
Zelinsky, A. A. Balandin, 4 pp.

RUSSIAN, per, Zbir Fil Khim, Vol. XXXV, No 3, 1961,
pp 481-489.

Cleaver-Hume Press Ltd.

Sci

Feb 62

182, 697

Kinetics of the Dehydrogenation of Alcohols on
Copper, by A. A. Balandin, P. Teteni, 6 pp.

RUSSIAN, per, Zhur Fiz Khim, Vol XXXV, No 1,
1961, pp 62-71.

Cleaver-Hume Press

Sci

177927

Dec 61

Balandin, A. A., Marushkin, M. N., and
Afanas'ev, M. M.
CONTACT DECOMPOSITION OF HYDROCARBONS:
CONVERSION OF NORMAL-BUTANE UNDER THE
ACTION OF CARBON DIOXIDE OVER NICKEL-
ALUMINA CATALYSTS. [1961] 4p. 25 refs.
Order from OTS or SLA \$1.10

61-16643

61-16643
I. Balandin, A. A.
II. Marushkin, M. N.
III. Afanas'ev, M. M.
IV. Title: Conversion...

Trans. of Acta Physicochimica U. R. S. S., 1942,
v. 17, p. 83-92.

DESCRIPTORS: *Hydrocarbons, *Butanes, Decomposition,
*Carbon dioxide, *Alumina-nickel catalysts,
Catalysis, Chemical reactions.

The conversion of n-butane and carbon dioxide over
nickel-alumina is investigated experimentally. A
scheme of the reaction mechanism is given which ac-
counts for a successive shortening of the carbon chain
and explains the formation of carbon, being in accord-
(Chemistry--Organic, TT, v. 6, no. 10) (over)

Office of Technical Services

61-18349

I. Balandin, A. A.
II. Marukyan, G. M.

Balandin, A. A. and Marukyan, G. M.
PRODUCTION OF α -METHYLSTYRENE BY CATALYTIC
DEHYDROGENATION OF ISOPROPYL-BENZENE

[1961] 3p. 2 refs.

Order from OTS or SLA \$1.10

61-18349

Trans. of [Akademiya Nauk SSSR]. Comptes Rendus
(Doklady) de l'Academie des Sciences de l'U. R. S. S.,
1945, v. 48, p. 482-483.

DESCRIPTORS: *Styrenes, *Methyl radicals, Synthesis,
*Cumenes, Dehydrogenation, *Chromium compounds,
*Oxides, Catalysts, Catalysis, Propyl radicals,
Benzenea.

Catalytic dehydrogenation of isopropylbenzene to
 α -methylstyrene on catalysts such as promoted chromium
oxide occurs at a higher rate than of ethylbenzene to
styrene under comparable conditions. The yield
(Chemistry-Organic, TT, v. 6, no. 10) (over)

Office of Technical Services

Balandin, A. A.
CATALYTIC DEHYDROGENATION OF HYDROCARBONS AND ITS APPLICATION TO SYNTHESIS OF RUBBER FROM GASES. [1961] 24p. 90 refs.
Order from OTS or SLA \$2.60

61-18058

Trans. of Akademiya Nauk SSSR. Otdelenie Khimicheskikh Nauk. Izvestiya, 1942, no. 1, p. 21-44.

DESCRIPTORS: *Hydrocarbons, *Synthetic rubber, *Dehydrogenation, *Catalysts, Benzenes, Butanes, Butenes, Synthesis, Butadienes, Styrenes, Ethyl radicals.

The catalytic dehydrogenation of hydrocarbons on chromic oxide is discussed from the point of view of the author's multiplet theory and examples of dehydrogenation of butane, butene and ethylbenzene given indicating the significance and the applicability of this method for commercial synthesis of butadiene and styrene. (Author)

61-18058

I. Balandin, A. A.

(Chemistry--Organic,
TT, v. 6, no. 10)
Office of Technical Services

Balandin, A. A., Marukyan, G. M., and
Sednovich, R. G.
CATALYTIC DEHYDROGENATION OF p-CYMENE.
[1961] 4p. 15 refs.
Order from CTS or SLA \$1.10

61-16920

Trans. of Akademii Nauk SSSR. Comptes Rendus
(Doklady) de l'Academie des Sciences de l'U. R. S. S.,
1949, v. 41, p. 71-73.

DESCRIPTORS: *Cymenes, Dehydrogenation,
Catalysis.

Introduction of two methyl groups into the molecule of
ethylbenzene, one attached to the ring and the other
placed in the side chain in a-position, facilitates de-
hydrogenation. (Author)

(Chemistry--Organic, TT, v. 6, no. 7)

61-16920

I. Balandin, A. A.
II. Marukyan, G. M.
III. Sednovich, R. G.

Office of Technical Services

Balandin, A. A. and Marushkin, M. N.
FORMATION OF OLEFINS FROM HIGHER PARAFFINS. [1961] 5p. 9 refs.
Order from OTS or SLA \$1.10

61-16928

Trans. of Akademiya Nauk SSSR. Comptes Rendus
(Doklady) de l'Academie des Sciences de l'U.R.S.S.
1943, v. 40, p. 254-256.

DESCRIPTORS: *Ethylenes, Synthesis, Hydrocarbons,
Decomposition, Waxes, Catalysis.

In a preliminary study catalytic cracking of paraffin wax in the presence of a mixed chromium catalyst gave results showing that at 450-500° dehydrogenation to olefins definitely predominates over other reactions of paraffin wax hydrocarbons under the conditions used.
(Author)

(Chemistry--Organic, TT, v. 6, no. 7)

61-16928

I. Balandin, A. A.
II. Marushkin, M. N

Office of Technical Services

Balandin, A. A., Zelinskii, N. D. and others.
PREPARATION OF 1,3-BUTADIENE BY CATALYTIC
DEHYDROGENATION OF 1-BUTENE. [1961] 9p.
15 refs.

Order from OTS or SLA \$1.10 61-18011

Trans. of Zhurnal Prikladnoi Khimii (USSR) 1941,
v. 14, p. 435-445.

Descriptors: *Butadienes, Synthesis, *Butenes,
Dehydrogenation, Catalysts.

The catalytic dehydrogenation of 1-butene to 1,3-butadiene was investigated in the presence of carbon dioxide or nitrogen. The conditions were established under which butadiene is formed in yields of up to 34% on the passed, or 77% on the decomposed, butene. The reaction is carried out under atmospheric pressure, at 600°, with a contact period of 0.3 seconds and dilution of butene with carbon dioxide in the ratio of 1:7.5 by

(Chemistry--Organic, TT, v. 6, no. 7) (over)

61-18011

I. Balandin, A. A.
II. Zelinskii, N. D.

Office of Technical Services

61-16636

I. Balandin, A. A.
ii. Zelinskii, N. D.

Balandin, A. A., Zelinskii, N. D. and others.
CATALYTIC DEHYDROGENATION OF BUTENE TO
BUTADIENE UNDER REDUCED PRESSURE. [1961]
11p. 15 refs.
Order from OTS or SLA \$1.60

61-16636

Trans. of Zhurnal Prikladnoi Khimii (USSR) 1942,
v. 15, p. 128-138.

DESCRIPTORS: *Butadienes, Dehydrogenation,
*Butenes, Chromic catalysts.

Research on dehydrogenation of 1-butene to 1,3-butadiene was continued by employing a pressure of 180 mm Hg and two chromium catalysts. The effect of the contact period and the temperature were systematically investigated. Occurrence of two consecutive reactions was established consisting in formation and decomposition of butadiene. The highest yield of butadiene was obtained at 59°C and a rate of flow of 2,000 liters per liter of catalyst per hour. This yield amounted to 29% (over:

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